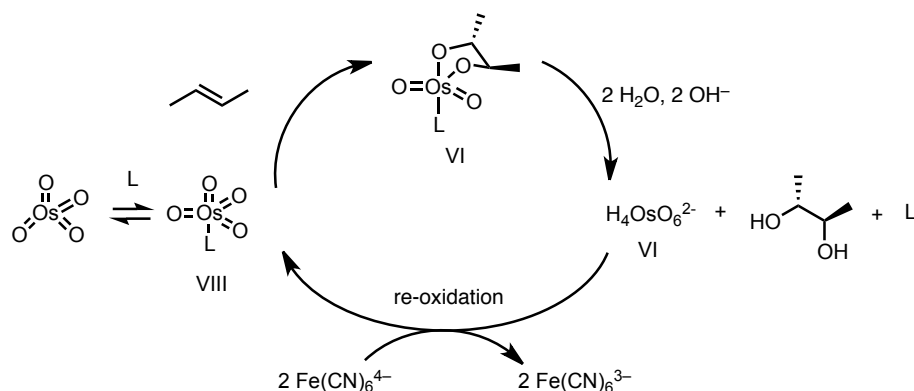


Reviews:

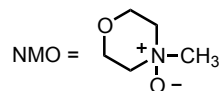
Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483–2547.
 Noe, M. C.; Letavic, M. A.; Snow, S. L.; McCombie, S. W. *Org. React.* **2005**, *66*, 109–626.

Ligands such as pyridine accelerate the osmylation of olefins (Criegee, R.; Marchand, B.; Wannowius, H. *Liebigs Ann. Chem.* **1942**, *550*, 99–133.)

Catalytic Cycle:



Turnover is achieved by reoxidation with stoichiometric oxidants:



UpJohn Process: VanRheenen, V.; Kelly, R. C.; Cha, D. Y. *Tetrahedron Lett.* **1976**, 1973–1976.

$\text{K}_3\text{Fe}(\text{CN})_6$ Minato, M.; Yamamoto, K.; Tsuji, J. *J. Org. Chem.* **1990**, *55*, 766–768.

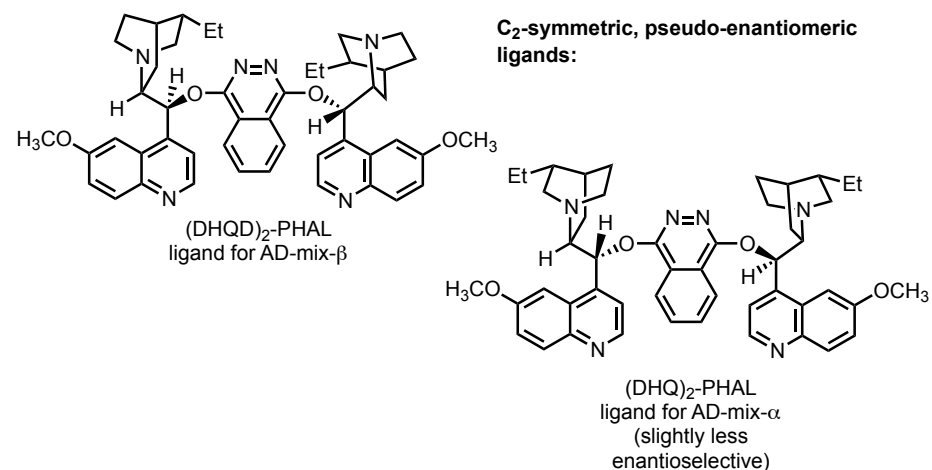
In the original Sharpless procedure using NMO, reoxidation was believed to compete with hydrolysis, leading to a ligand-less "second cycle" olefin dihydroxylation that was non-enantioselective:

Ogino, Y.; Chen, H.; Kwong, H.-L.; Sharpless, K. B. *Tetrahedron Lett.* **1991**, *32*, 3965–3968.

Balance of evidence favors 3+2 cycloaddition (vs. 2+2/rearrangement) mechanism between osmium and olefin.

See, e.g., Corey, E. J.; Noe, M. C.; Grogan, M. J. *Tetrahedron Lett.* **1996**, *37*, 4899–4902.

DelMonte, A. J.; Haller, J.; Houk, K. N.; Sharpless, K. B.; Singleton, D. A.; Strassner, T.; Thomas, A. A. *J. Am. Chem. Soc.* **1997**, *119*, 9907–9908.



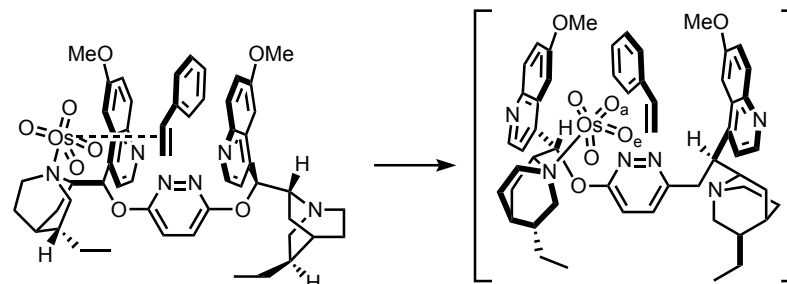
AD-mix reagents are commercially available:

1.4 g AD-mix-β will oxidize 1 mmol olefin, contains:
 0.98 g $\text{K}_3\text{Fe}(\text{CN})_6$ (3 mmol)
 0.41 g K_2CO_3 (3 mmol)
 0.0078 g (DHQD)₂-PHAL (0.01 mmol)
 0.00074 g $\text{K}_2\text{OsO}_2(\text{OH})_4$ (0.002 mmol)

Conditions: *t*-BuOH, H₂O (1:1), 0 °C, 6–24 h
 Typical work-up: Na_2SO_3 then extraction

Sharpless, K. B., et al. *J. Org. Chem.* **1992**, *57*, 2768–2771.

Corey proposes a U-shaped binding pocket:

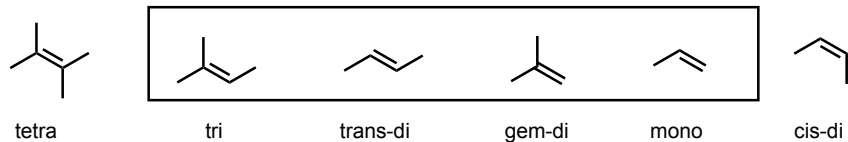


Corey, E. J.; Guzman-Perez, A.; Noe, M. C. *Tetrahedron Lett.* **1995**, *36*, 3481–3484.

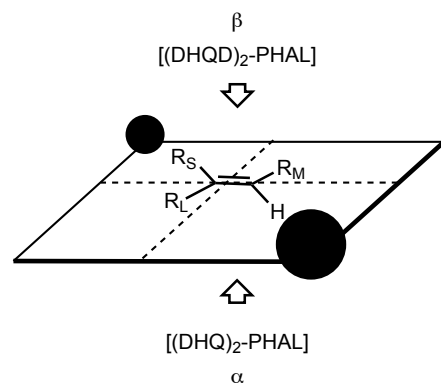
For ligand modifications and improvements based on binding model, see:

Corey, E. J.; Noe, M. C.; Grogan, M. J. *Tetrahedron Lett.* **1994**, *35*, 6427–6430.
 Huang, J.; Corey, E. J. *Org. Lett.* **2003**, *5*, 3455–3458.

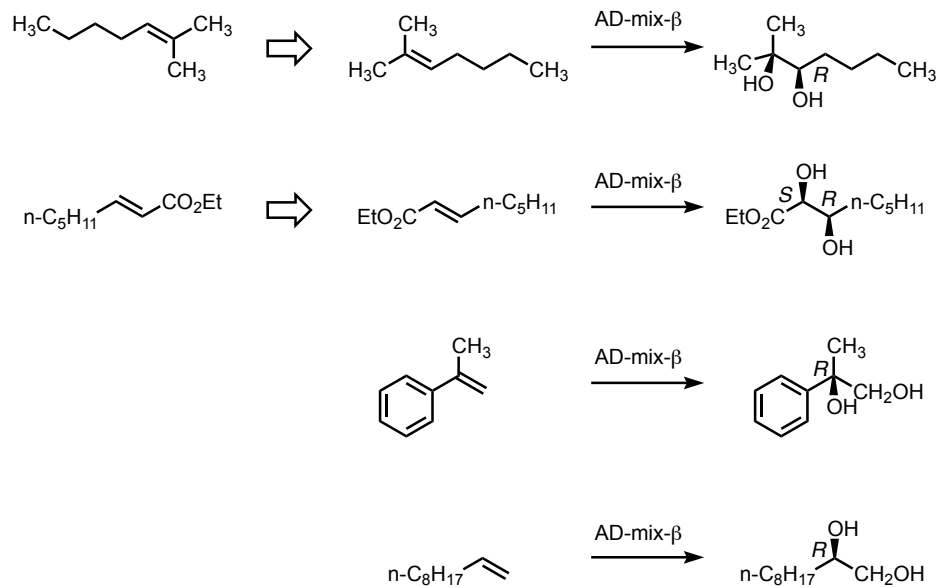
4 of 6 Olefin substitution classes are successfully dihydroxylated:



Mnemonic:



Application of Mnemonic:

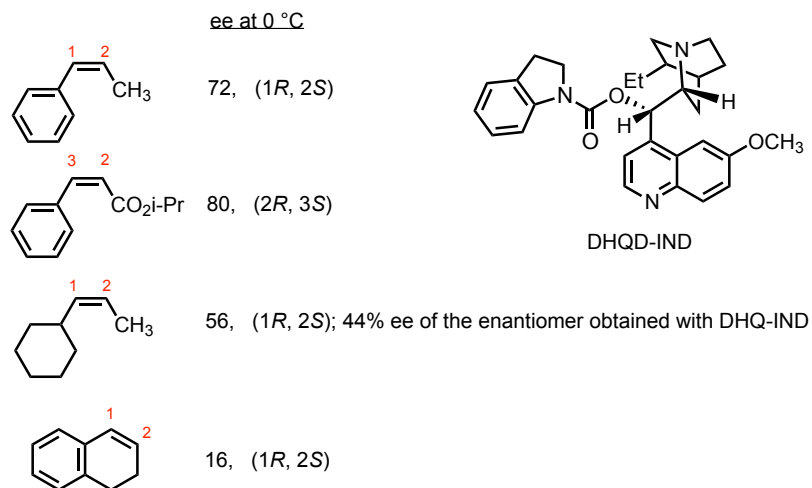


	AD-mix- β [(DHQD) ₂ -PHAL] % ee, config.	AD-mix- α [(DHQ) ₂ -PHAL] % ee, config.
$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}=\text{C}(\text{CH}_3)_2$ *	98, <i>R</i>	95, <i>S</i>
$\text{Ph}-\text{CH}=\text{CH}-\text{Ph}$ *	99, <i>R, R</i>	97, <i>S, S</i>
$\text{n-Bu}-\text{CH}=\text{CH}-\text{n-Bu}$ *	97, <i>R, R</i>	93, <i>S, S</i>
$\text{n-C}_5\text{H}_{11}-\text{CH}=\text{CH}-\text{CO}_2\text{Et}$ *	99, 2 <i>S</i> , 3 <i>R</i>	96, 2 <i>R</i> , 3 <i>S</i>
$\text{Ph}-\text{CH}=\text{CH}-\text{CO}_2\text{Et}$ *	97, 2 <i>S</i> , 3 <i>R</i>	95, 2 <i>R</i> , 3 <i>S</i>
$\text{Ph}-\text{CH}=\text{CH}-\text{Ph}$ *	>99.5, <i>R, R</i>	>99.5, <i>S, S</i>
$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}=\text{C}(\text{CH}_3)_2$	78, <i>R</i>	76, <i>S</i>
$\text{Ph}-\text{CH}=\text{CH}_2$	94, <i>R</i>	93, <i>S</i>
$\text{n-C}_8\text{H}_{17}-\text{CH}=\text{CH}_2$	84, <i>R</i>	80, <i>S</i>
$\text{Ph}-\text{CH}=\text{CH}_2$	97, <i>R</i>	97, <i>S</i>

* addition of $\text{CH}_3\text{SO}_2\text{NH}_2$ (a phase-transfer and general acid catalyst) leads to faster reactions for non-terminal olefins

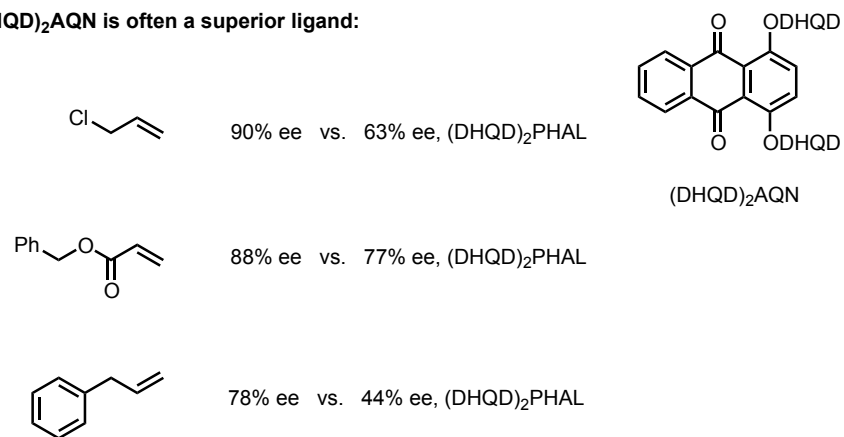
Sharpless, K. B., et al. *J. Org. Chem.* **1992**, 57, 2768–2771.

Cis-disubstituted olefins are generally poor substrates. With a modified catalyst, DHQD-IND, fair to good enantioselectivities can be obtained:



Wang, L.; Sharpless, K. B. *J. Am. Chem. Soc.* **1992**, *114*, 7568–7570.

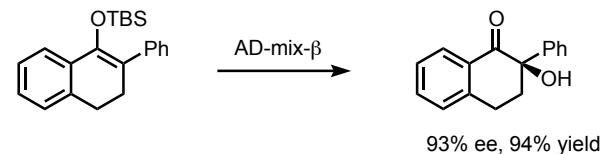
(DHQD)₂AQN is often a superior ligand:



Becker, H.; Sharpless, K. B. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 448–451.

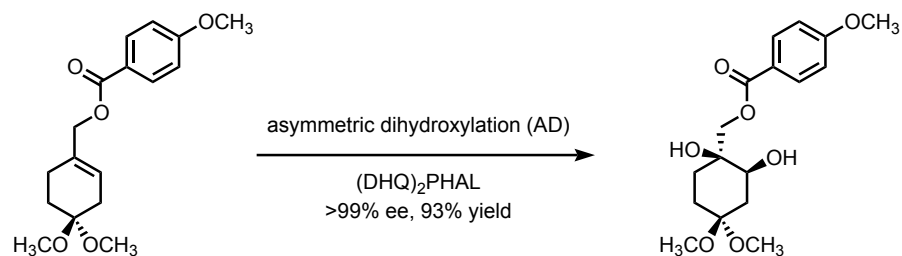
A few tetra-substituted olefins work well:

Sharpless, K. B., et al. *J. Am. Chem. Soc.* **1993**, *115*, 8463-8464

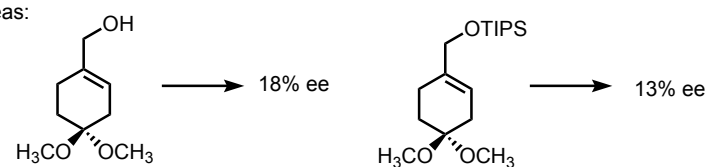


a best case; ee's and yields are not generally high

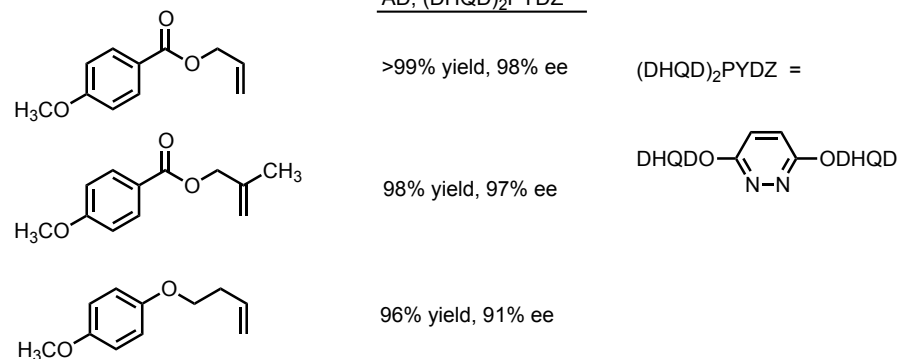
Allylic 4-methoxybenzoates are particularly good substrates:



whereas:

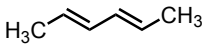
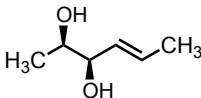
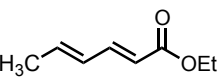
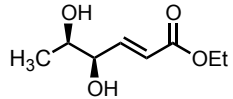
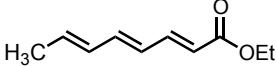
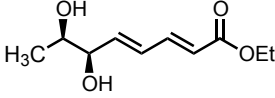
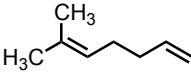
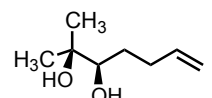
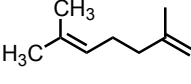
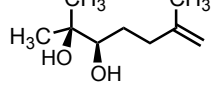


AD, (DHQD)₂PYDZ



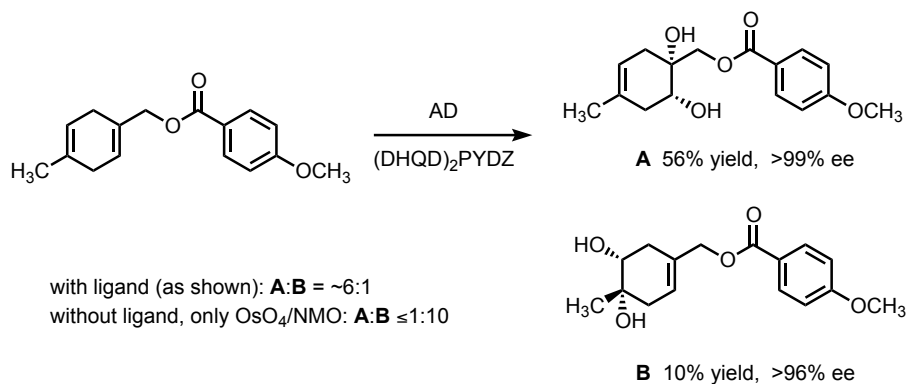
Corey, E. J.; Guzman-Perez, A.; Noe, M. C. *J. Am. Chem. Soc.* **1995**, *117*, 10805–10816.

Regioselectivity of AD with diene substrates ((DHQD)₂PHAL as ligand):

Substrate	Product	% yield, % ee
		78, 93
		78, 92
		93, 95
		73, 98
		70, 98

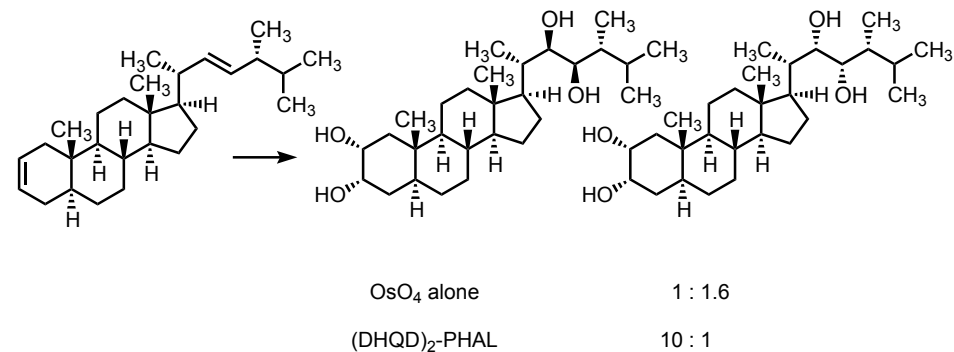
in general, AD is selective for the more electron-rich double bond

Xu, D.; Crispino, G. A.; Sharpless, K. B. *J. Am. Chem. Soc.* **1992**, *114*, 7570-7571.

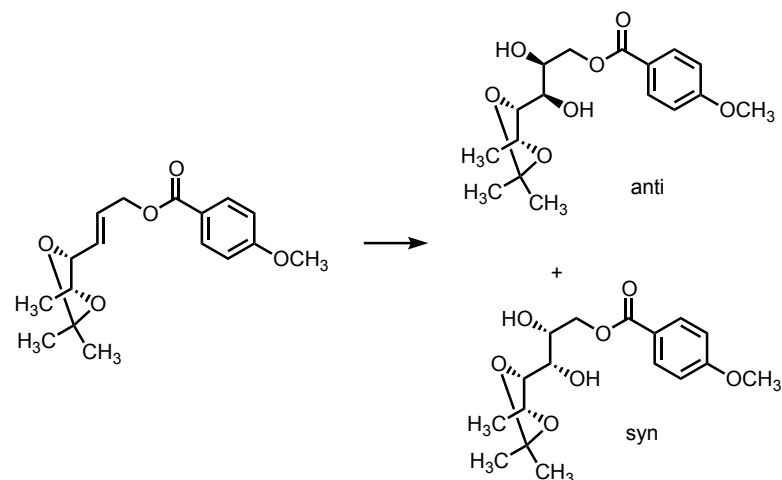


Corey, E. J.; Guzman-Perez, A.; Noe, M. C. *J. Am. Chem. Soc.* **1995**, *117*, 10805-10816.

Use of AD with chiral olefins:



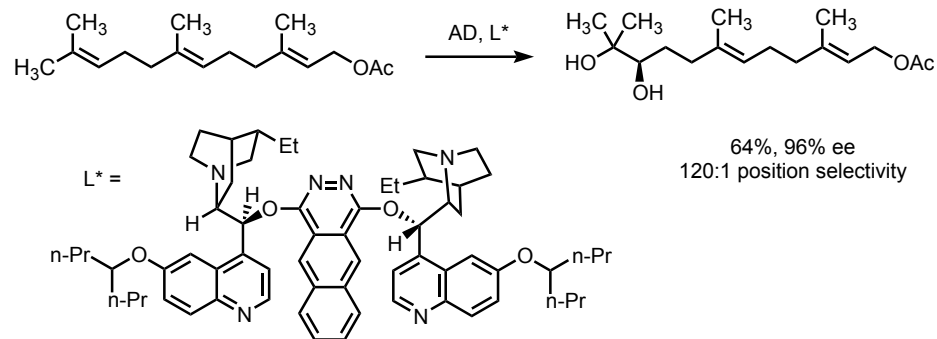
Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. *Chem. Rev.* **1994**, *94*, 2483-2547, and refs. therein.



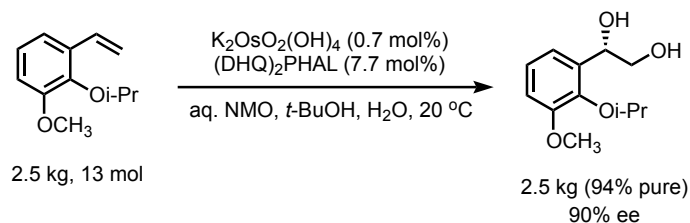
Conditions	Yield	anti : syn
OsO ₄ , NMO	88% yield (mixture)	1.9 : 1
(DHQ) ₂ PHAL (matched)	86% yield (anti)	54 : 1
(DHQD) ₂ PYDZ (mismatched)	86% yield (syn)	1 : 35

Guzman-Perez, A.; Corey, E. J. *Tetrahedron Lett.* **1997**, *38*, 5941-5944.

Regioselective AD of terminal olefin of oligoprenyl derivatives:

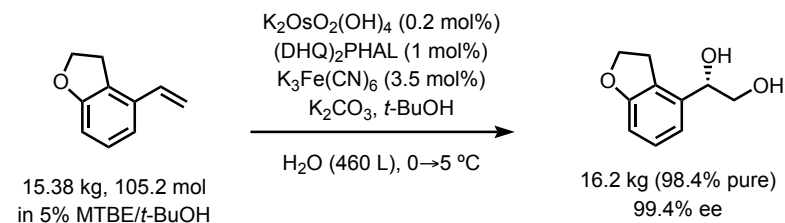


Examples in Industry



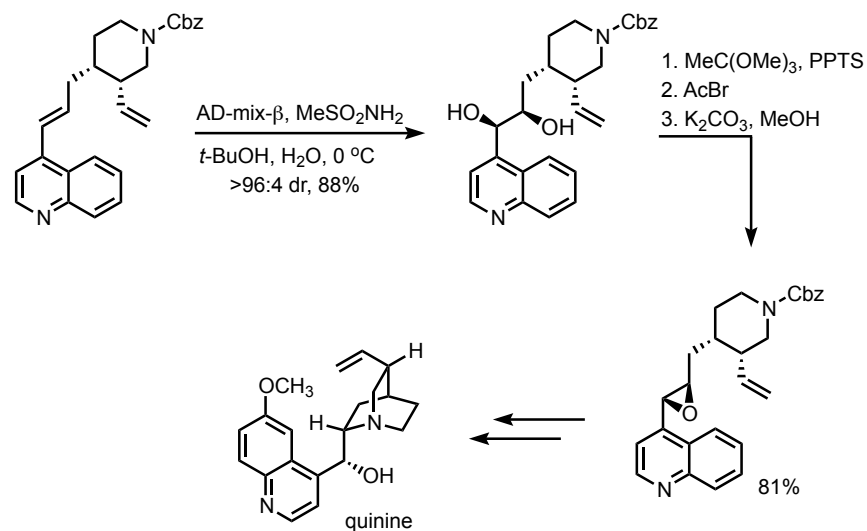
• Olefin was added over a period of 6.5 h to the reaction mixture to prevent "second cycle" oxidation.

Ahrgren, L.; Sutin, L. *Org. Process Res. Dev.* **1997**, *1*, 425–427.



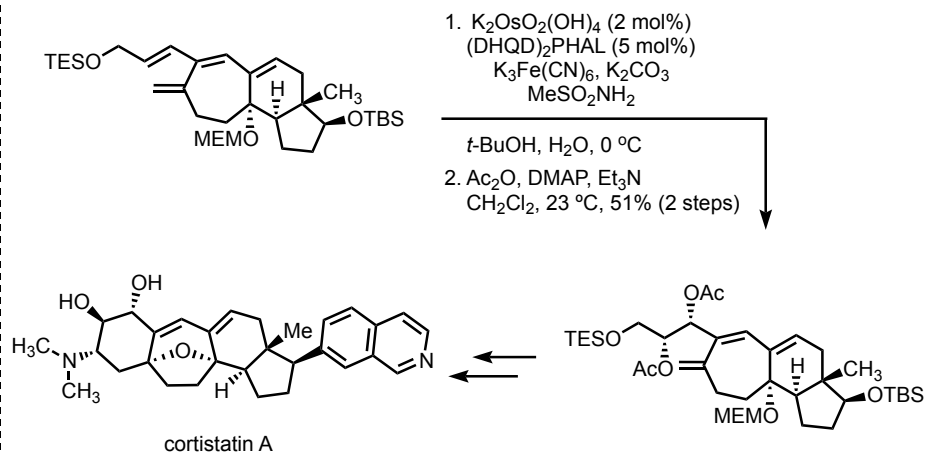
Prasad, J. S.; Vu, T.; Tottleben, M. J.; Crispino, G. A.; Kacsur, D. J.; Swaminathan, S.; Thornton, J. E.; Fritz, A.; Singh, A. K. *Org. Process Res. Dev.* **2003**, *7*, 821–827.

Examples in Syntheses of Natural Products:



Raheem, I. T.; Goodman, S. N.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2004**, *126*, 706–707.

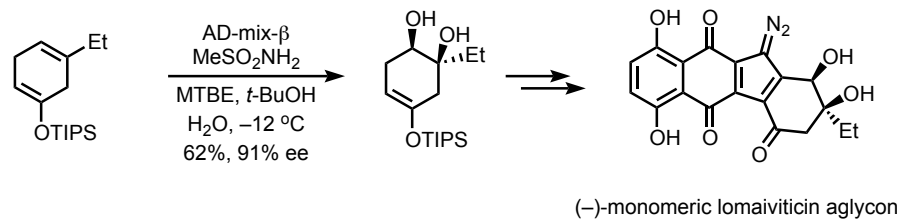
For conversion of diol to epoxide, see Kolb, H. C.; Sharpless, K. B. *Tetrahedron* **1992**, *48*, 10515–10530.



Lee, H. M.; Nieto-Oberhuber, C.; Shair, M. D. *J. Am. Chem. Soc.* **2008**, *130*, 16864–16866.

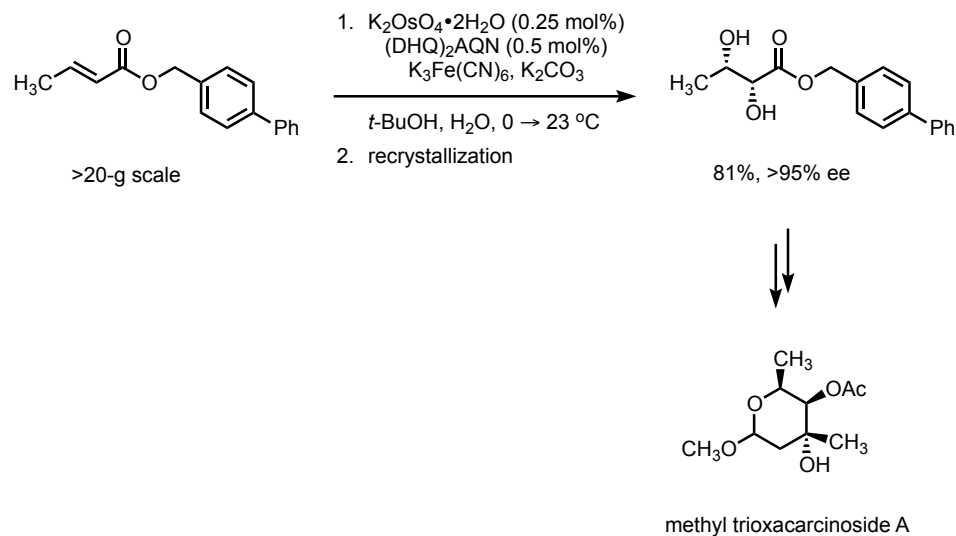
Adam Kamlet

- In the example below, use of the triisopropylsilyl protecting group was crucial to achieve regioselectivity:



Woo, C. M.; Gholap, S. L.; Lu, L.; Kaneko, M; Li, Z.; Ravikumar, P. C.; Herzon, S. B. *J. Am. Chem. Soc.* **2012**, *134*, 17262–17273.

- In the example below, a 4-phenylbenzyl ester was incorporated to serve as an expedient for purification and enantioenrichment by re-crystallization:



Smaltz, D. J.; Myers, A. G. *J. Org. Chem.* **2011**, *76*, 8554–8559.

Smaltz, D. J.; Svenda, J.; Myers, A. G.; *Org. Lett.* **2012**, *14*, 1812–1815.